

## Thallium Interaction with the Gastric (K, H)-ATPase

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**Summary.** The gastric (K,H)-ATPase has been shown to catalyze an electroneutral  $H^+$  for  $K^+$  exchange.  $Tl^+$  is able to substitute for  $K^+$  as an activating cation in the hydrolytic reaction with an apparent dissociation constant of  $90 \mu M$  as compared to about  $870 \mu M$  for  $K^+$ . The ability of  $Tl^+$  to participate in transport is shown by the development of pH gradients in the presence of  $Tl^+$  following addition of ATP to gastric vesicles and by the ATP-dependent efflux of  $Tl^+$  from gastric vesicles. Inhibition of hydrolysis is observed at pH 7.4 with external  $Tl^+$  concentrations above  $3.0 mM$ . This inhibition of hydrolysis is correlated with inhibition of pH-gradient formation. The inhibition of transport activity is partially relieved by a decrease in medium pH. This inhibitory effect is attributed to  $Tl^+$  binding at an external, low affinity cation site. In contrast to rubidium chloride, at high  $Tl^+$  concentrations, following the initial  $Tl^+$  efflux, there is reuptake of the cation. This rapid uptake is attributed to lipid-dependent  $Tl^+$  entry pathways. The vesicles exhibit a high permeability to thallium nitrate demonstrating a half-time ( $t_{1/2}$ ) for uptake of about 1.0 min in contrast to 46 min for rubidium chloride. In both gastric vesicles or liposomes, external  $Tl^+$  concentrations in excess of 1 to 4 mM are able to dissipate intravesicular proton gradients by an electrically coupled  $H^+$  for  $Tl^+$  exchange. Thus, although  $Tl^+$  is able to activate the gastric ATPase by mimicking  $K^+$ , the permeability of this cation in lipid bilayers tends to uncouple  $H^+$  transport at concentrations high enough to generate detectable proton gradients.

**Key words:** Thallium transport, gastric (K,H)-ATPase, cation activation site, cation inhibitory site

$Tl^+$  is a group III cation with at least two types of interaction with biological membranes that have been

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examined in recent years. On one hand, it has been shown that  $Tl^+$  can act as a lipid-permeable cation in mitochondria (Melnick, Monti & Motzkin, 1976; Skulskii, Savina, Glasunov & Saris, 1978) or starved, valinomycin-treated cells of streptococcus lactis 7962 (Kashket, 1979), distributing across the membrane at the same ratio as other lipid-permeable cations such as tetraphenylphosphonium. It also seems to behave as such a cation in red cells distributing in a ratio opposite to that of  $Cl^-$  (Skulskii, Manninen & Jarnefelt, 1973), which is in electrochemical equilibrium with the measured membrane potential. It has also been recognized that this cation interacts with the (Na,K)-ATPase of erythrocytes (Skulskii et al. 1973) and is transported by this enzyme in squid axon (Landowne, 1975). Presumably, therefore, in red cells the flux of  $Tl^+$  via lipid pathways greatly exceeds the small pump capacity in this cell. Attempts to use this cation for measuring membrane potentials in bacteria were frustrated by the finding that bacteria contain  $K^+$  transport systems which also transport  $Tl^+$  (Bakker, 1978; Bakker-Grunwald, 1979; Kashket, 1979).

The gastric mucosal ATPase (Ganser & Forte, 1973b) catalyzes an electroneutral exchange of  $H^+$  for  $K^+$  with accumulation of  $H^+$  in the vesicle lumen and transport of  $K^+$  to vesicle exterior. The rate of transport is determined in part by  $K^+$  binding to an internal high-affinity site which is required for activation of ATP hydrolysis and in part by  $K^+$  binding to an external low-affinity site which slows hydrolysis and transport (Stewart, Wallmark & Sachs, 1980; Wallmark, Stewart, Rabon, Saccomani & Sachs, 1980). The relative affinity of  $Tl^+$  (Ray & Forte, 1976) for participation in the  $K^+$ -dependent reactions of the (K,H)-ATPase may provide a clear distinction between  $K^+$  activation and inhibitory sites.

The ability of  $Tl^+$  to act as a lipid-permeable cation may also prove useful in determining the electrogenic

characteristics of the pump as well as helping to define the nature of the proton leak pathways in gastric vesicles.

## Materials and Methods

### *Preparation of Gastric Membranes*

The method used has been detailed previously. Briefly, the hog gastric mucosal epithelial surface was flooded with 3 M NaCl and incubated for 2 min. The NaCl solution and the majority of the surface epithelial cells were removed by wiping the surface with absorbant tissue. The remaining tissue was then scraped with a blunt spatula and the scrapings then homogenized in 0.25 M sucrose. Following differential centrifugations the microsomal pellet was centrifuged through an interface between 0.25 M sucrose and 7% Ficoll in 0.25 M sucrose. The material designated GI, that trapped at the interface, was used throughout these experiments (Sachs, Chang, Rabon, Schackmann, Lewin & Saccomani, 1976).

### *ATPase Activity*

This was measured in a final volume of 1.0 ml containing 2 mM MgNO<sub>3</sub>, 40 mM Tris(hydroxymethyl)aminomethane (TrisNO<sub>3</sub>), pH 7.4, 10 µg enzyme protein and TiNO<sub>3</sub> between 0.001 and 50 mM and/or KNO<sub>3</sub> between 0.5 mM and 20 mM. On occasion, as detailed in the text, Tl<sup>+</sup> and K<sup>+</sup> were present simultaneously. The effect of ionophores on cation-activated ATPase activity was assessed with either gramicidin or nigericin added in 10 µl methanol to give a final concentration of 1 µg/ml of ionophore. Controls included in the assay indicate that a 10 µl addition of methanol did not appreciably stimulate the cation-stimulated component of ATP hydrolysis. The release of Pi was measured after 15 min at 37 °C by the method of Yoda and Hokin (1970) or by radiometric assay. Tl<sup>+</sup> above 3.0 mM interferes with the colorimetric assay of Pi release. <sup>γ</sup><sup>32</sup>P-ATP was used to measure enzyme activity when Tl<sup>+</sup> concentrations were higher than 3.0 mM. Here a 300 µl assay medium contained 40 mM (Piperazine-N, N'-bis{2-ethane sulfonic acid}) (PIPES/Tris) at pH 7.4, 2.0 mM MgNO<sub>3</sub>, 2.0 mM <sup>γ</sup><sup>32</sup>P-ATP (≈ 1000 cpm/nmole), 5.0 µg protein and TiNO<sub>3</sub> within the concentration range 6.0 to 50 mM. After a 5-min incubation at 37 °C enzyme activity was stopped by the addition of 17 mM CDTA to the samples which were immediately placed on ice. The samples were diluted to 1.0 ml with ice-cold distilled water. A second volume of 1.0 ml which contained 5% ammonium molybdate in 4.0 N HNO<sub>3</sub> was then added and the resulting complex extracted into 2.5 ml of 1:1 isobutanol-benzene. ATPase activity was calculated from the <sup>32</sup>Pi partitioning into a 1.0 ml aliquot of the organic layer. Protein was measured by the Lowry method (Lowry, Rosebrough, Farr & Randall, 1951).

### *<sup>204</sup>Tl<sup>+</sup> Trapping*

Gastric vesicles at 0.5 mg/ml were added to an incubation solution containing (in mM): 150 KNO<sub>3</sub>, 2.0 MgNO<sub>3</sub>, 2.0 PIPES/Tris, pH 7.4, and 6.0 <sup>204</sup>Tl<sup>+</sup> at a final activity of 0.011 µCi/µl. At different times following this addition, aliquots were taken from the mixture, filtered on HAWP Millipore filters and washed 4 times with ice-cold 276.0 mM mannitol, 2.0 mM MgNO<sub>2</sub> and 2.0 mM PIPES/Tris, pH 7.4. The filters were dried and counted in an LKB 81000 liquid scintillation counter.

### *ATP-Dependent Tl<sup>+</sup> Transport*

The gastric vesicles at 0.5 mg ml<sup>-1</sup> were preincubated for 2 hr at room temperature in a medium containing (in mM): 2.0 MgNO<sub>3</sub>,

2.0 PIPES/Tris, pH 7.4, 150 KNO<sub>3</sub> and either 0.6 or 6.0 <sup>204</sup>Tl<sup>+</sup> to give a radioactivity of 0.011 µCi/µl<sup>-1</sup>. In addition to its contribution to pump activity K<sup>+</sup> contributed to the maintenance of lower Tl<sup>+</sup> blanks which can be problematic in this assay. Transport was initiated by adding to 100 µl of this suspension, 5 µl of a solution containing 42 mM ATP, 42 mM MgNO<sub>3</sub>, 50 mM PIPES/Tris buffer, pH 7.4. At various times, the reaction mixture was filtered on a 0.45 µ Millipore HAWP filter and washed 4 times with a solution containing 276 mM mannitol, 2 mM MgNO<sub>3</sub> and 2 mM PIPES/Tris, pH 7.4. The dried filters were counted in 10 ml of ACS scintillation fluid in an LKB 81000 scintillation counter.

### *Incubation Procedure for Substrate-Dependent Transport and Experiments Involving pH Pulses*

For studies of ATP-dependent proton transport measured by the acridine orange absorbance technique the gastric vesicles were preincubated for 2 hr at room temperature in a medium containing at final concentration 150 mM KNO<sub>3</sub>, 2 mM MgNO<sub>3</sub>, 2 mM PIPES/Tris, pH 7.4. For studies using the pH pulse technique 5.0 mM succinate/Tris, pH 5.5, was substituted for the PIPES/Tris buffer usually used. When a pH electrode was used to measure proton transport the pH was adjusted to 6.1 using 1.0 mM PIPES/Tris, pH 6.13 (Sachs et al., 1976).

### *Measurement of ATP-Dependent pH-Gradient Formation*

Using the vesicles preincubated in KNO<sub>3</sub> media as above, the formation of a pH gradient was monitored following the addition of ATP by the time-dependent change in acridine orange absorbance. In previous experiments, the loss of the absorbance peak at 490 nm was correlated with acidification of the intravesicular space (Rabon, Chang & Sachs, 1978). This peak was measured in an Aminco DW-2 Spectrophotometer in the dual wavelength mode with the monochromators set at 490 and 446 nm. For this assay 0.8 mM MgATP was added to a 350-µl assay solution containing 150 mM KNO<sub>3</sub>, 2.0 mM MgNO<sub>3</sub>, 2.0 mM PIPES/Tris (pH 6.1 or pH 7.4), 15 µM acridine orange and 15 µg of membrane protein. When pH electrode measurements were used 0.3 mM MgATP was added to a 2.0-ml assay solution as described above buffered at pH 6.1 and containing 0.33 mg protein. Changes in external pH following substrate addition were measured using a Radiometer PHM 64 pH meter and an REA 12 high-sensitivity unit interfaced with the REC-61 Servorecorder. Signal calibration was as previously detailed (Rabon et al., 1978).

### *Formation of Liposomes*

Thirty-five mg of L-α-phosphatidyl choline was sonicated for 40 min in 1.0 ml of a solution consisting of 276 mM mannitol, 2.0 mM MgNO<sub>3</sub>, 5 mM succinate/Tris, pH 5.5, using a Laboratory Supplies Co., Inc. Sonicator at 80 W output. The liposomes were stored at 4 °C and used within 48 hr of preparation.

### *Generation of a pH Gradient*

In the case of liposomes formed as above, 10 µl of liposomal suspension was added to a 340 µl solution containing 276 mM mannitol, 2.0 mM MgNO<sub>3</sub>, 15 µM acridine orange and 1.0 mM succinate/Tris, pH 5.5. The pH gradient was induced by addition of 100 µl of 100 mM Tris/NO<sub>3</sub>, pH 8.5, 176 mM mannitol, 15 µM acridine orange and 2.0 mM MgNO<sub>3</sub>. For the gastric vesicles, 60 µl of the vesicles, equilibrated in KNO<sub>3</sub> medium as above, were diluted in 290 µl of medium containing 150 mM KNO<sub>3</sub>, 2.0 mM MgNO<sub>3</sub>,

15  $\mu\text{M}$  acridine orange and 1.0 mM succinate/Tris, pH 5.5. The pH gradient was formed by addition of 100  $\mu\text{l}$  of solution containing 150 mM Tris/ $\text{NO}_3^-$ , pH 8.5, 56 mM  $\text{KNO}_3$ , 2.0 mM  $\text{MgNO}_3$  and 15  $\mu\text{M}$  acridine orange. In the gastric vesicles 25- $\mu\text{l}$  aliquots of  $\text{Tl}^+$  at varying concentrations were added 35 sec after the pH pulse. Because of the leakiness of the liposomal vesicles to protons, a 25- $\mu\text{l}$  aliquot of  $\text{Tl}^+$  was added to the alkaline medium before its addition to the liposome suspension.

### Materials

L- $\alpha$ -phosphatidylcholine and ATP were obtained from Sigma Chemical Co.,  $^{204}\text{Tl}^+$  from Amersham, acridine orange from Eastman Kodak and Ficoll from Pharmacia. All other chemicals were reagent grade or better.

### Results

#### $\text{Tl}^+$ -Dependent Activation of ATPase

The comparison of nonionophoric stimulation of hydrolysis by the (K,H)-ATPase by  $\text{Tl}^+$  in intact gastric vesicles to the ionophoric stimulation is shown in Fig. 1. As shown, there is a significant gramicidin-stimulated component of hydrolysis up to 3.0 mM  $\text{Tl}^+$ . The insert of Fig. 1 is another experiment which uses a Lineweaver-Burk plot to compare the gramicidin-stimulated ATPase activity dependent upon  $\text{Tl}^+$  (0.05 to 3.0 mM) to that of  $\text{K}^+$  (0.5 to 20 mM). In this experiment, calculation of the apparent affinity of the cation activation site for these cations indicates that the affinity for  $\text{Tl}^+$  is approximately 10-fold that of  $\text{K}^+$ . The apparent  $K_{0.5}$  is 89  $\mu\text{M}$  for  $\text{Tl}^+$  and 870  $\mu\text{M}$  for  $\text{K}^+$ . The  $V_{\text{max}}$  is approximately the same for each cation. Some experimental variability is noted in absolute activity between experiments. For example in 5 separate assays the  $K_{0.5}$  for  $\text{Tl}^+$  was determined to be  $0.098 \pm 0.013$  mM while  $V_{\text{max}}$  was  $129.8 \pm 13.4$   $\mu\text{mole mg}^{-1} \text{hr}^{-1}$ . This variability does not affect the affinity difference noted between  $\text{K}^+$  and  $\text{Tl}^+$ .

The addition of 20 mM  $\text{K}^+$  and 1  $\mu\text{g}$  of gramicidin per 10  $\mu\text{g}$  protein is routinely used to produce maximal stimulation of hydrolysis in the gastric vesicles (Ganser & Forte, 1973a; Sachs, 1977). The addition of 1.0 mM  $\text{Tl}^+$  to the assay medium containing 20 mM  $\text{K}^+$  and gramicidin did not produce an additional stimulation. In Fig. 2,  $\text{Tl}^+$  concentrations above 3.0 mM are shown to progressively inhibit maximal ATPase activity, as has been shown for  $\text{K}^+$  under these conditions (Wallmark et al., 1980). At 50 mM  $\text{Tl}^+$  only 15% of the maximal gramicidin-stimulated activity is present. At pH 7.4 and 37  $^\circ\text{C}$  the apparent affinity of this cation for the inhibitory site is at least 2 orders of magnitude less than that seen for activation. It must be noted that these assays are not directly comparable since conditions were not adjusted to

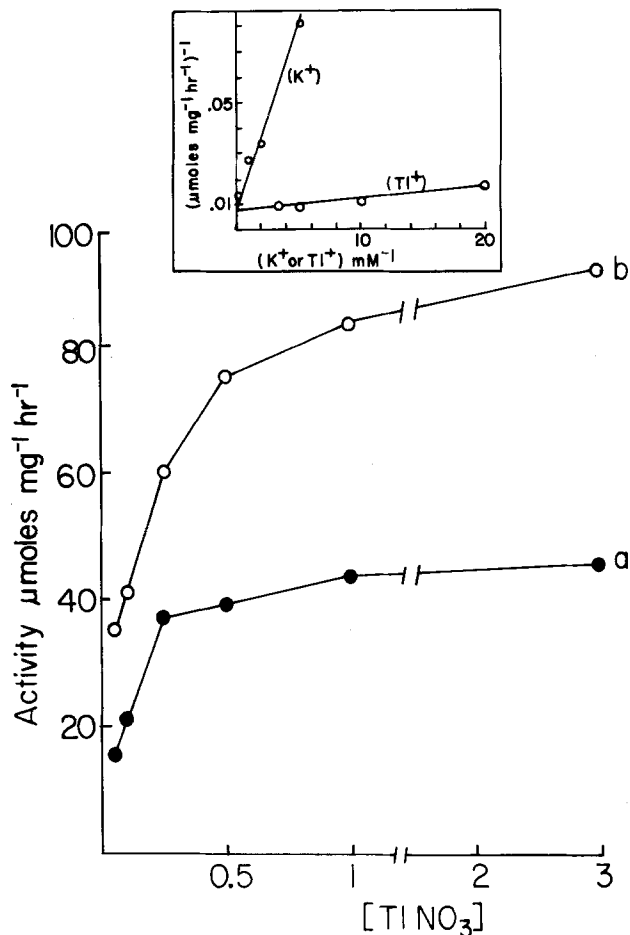


Fig. 1.  $\text{Tl}^+$ -dependent activation of the (K,H)-ATPase in gastric vesicles. Gastric vesicles containing 10  $\mu\text{g}$  protein were incubated in  $\text{Tl}^+$  concentrations from 50  $\mu\text{M}$  to 3.0 mM for 15 min at 37  $^\circ\text{C}$  in the assay medium listed in Materials and Methods. Trace a indicates the  $\text{Tl}^+$ -activation in the absence of gramicidin. Trace b indicates the  $\text{Tl}^+$ -activation in the presence of gramicidin. Basal  $\text{Mg}^{++}$ -ATPase activity is subtracted in each case. The insert is a Lineweaver-Burk plot of the gramicidin-stimulated hydrolysis in nonidentical concentrations of either  $\text{Tl}^+$  or  $\text{K}^+$

maintain constant ionic strength or the time of incubation between these assays.

#### $^{204}\text{Tl}^+$ Uptake by Gastric Vesicles

$^{204}\text{Tl}^+$  is rapidly accumulated by gastric vesicles as shown in Fig. 3. It can be seen that the  $t_{1/2}$  at room temperature is approximately 1.0 min, which can be compared to the  $t_{1/2}$  of 46 min observed for  $^{86}\text{Rb}^+$  (Schackmann, Schwartz, Saccomani & Sachs, 1977). Since it has been shown in mitochondria (Skulskii et al., 1978) that  $\text{Tl}^+$  distributes electrophoretically in accord with the membrane potential, the rapid uptake in gastric vesicles is presumably due to the lipid permeability of this cation. The equilibrium value for  $^{205}\text{Tl}^+$  uptake gives a calculated vesicular volume of 3.4  $\mu\text{l mg}^{-1}$  protein. This is about 1.5 times

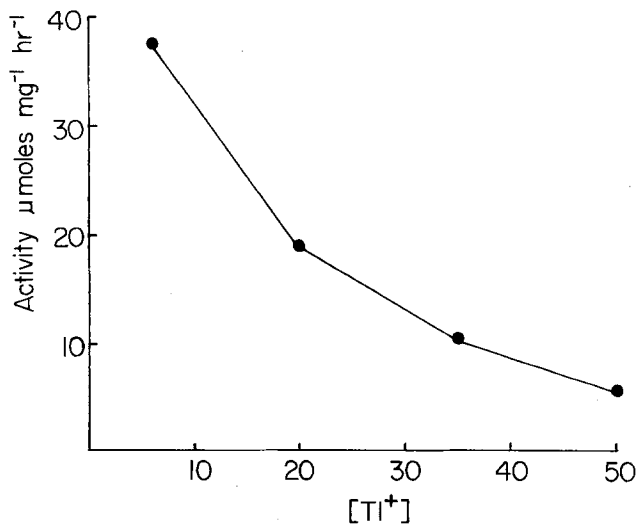


Fig. 2.  $\text{TI}^+$ -dependent inhibition of the (K,H)-ATPase in gastric vesicles. ATP hydrolysis was measured as indicated in Materials and Methods using  $2.0 \text{ mM } \gamma^{32}\text{P-ATP}$  in a reaction medium containing from  $6.0$  to  $50 \text{ mM } \text{TI}^+$  buffered at pH 7.4 and thermostated at  $37^\circ\text{C}$ . Activity was measured in the presence of gramicidin

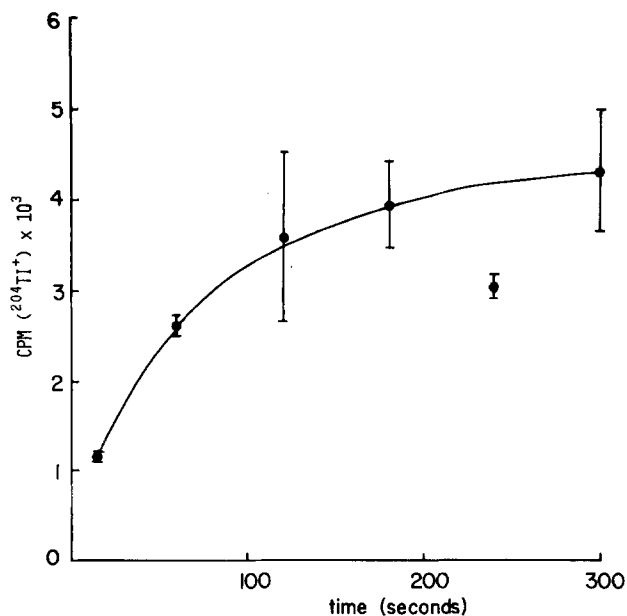


Fig. 3. Uptake of  $^{204}\text{TI}^+$  by gastric vesicles. The time-dependent uptake of  $^{204}\text{TI}^+$  was measured after the addition of the gastric vesicles to a medium containing (in mM):  $150 \text{ KNO}_3$ ,  $2.0 \text{ MgNO}_3$ ,  $10 \text{ Tris NO}_3$ , pH 7.4, and  $6.0 \text{ }^{204}\text{TI}^+$ . At the indicated times a  $100\text{-}\mu\text{l}$  aliquot containing  $50 \mu\text{g}$  protein was removed from the medium, filtered and washed on a HAWP Millipore filter. Each point represents the average of duplicate aliquots

the volume found with other cations, such as rubidium. This might be taken to imply binding of  $\text{TI}^+$ .

#### ATP-Dependent $^{204}\text{TI}^+$ Distribution

As discussed below, progressive inhibition of substrate-dependent proton transport is produced by the

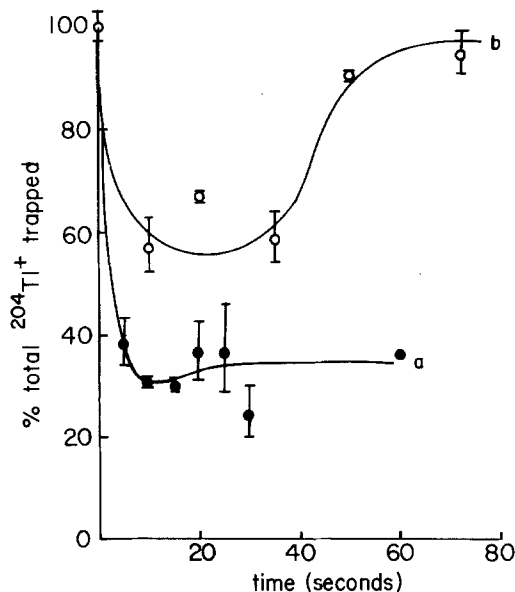


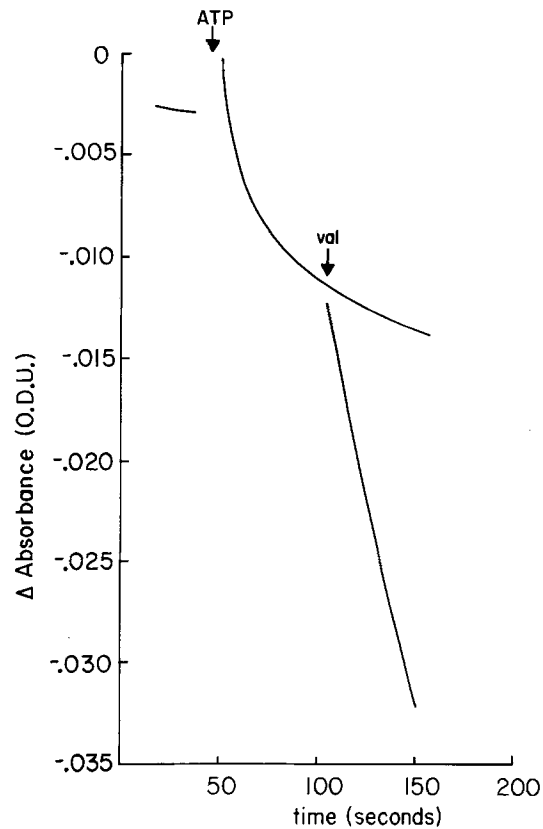
Fig. 4. ATP-dependent loss of equilibrated  $^{204}\text{TI}^+$  from gastric vesicles. Gastric vesicles were equilibrated in  $150 \text{ mM } \text{KNO}_3$  and  $0.6$  or  $6.0 \text{ mM } ^{204}\text{TI}^+$  medium as outlined in Materials and Methods. Following the addition of  $2.0 \text{ mM } \text{MgATP}$ , aliquots of the suspension were filtered on Millipore filters, washed and subsequently counted. In trace *a* the equilibrated  $^{204}\text{TI}^+$  concentration was  $0.6 \text{ mM}$ . In trace *b* the equilibrated  $^{204}\text{TI}^+$  concentration was  $6.0 \text{ mM}$ . Points are the average of duplicate assays

addition of  $\text{TI}^+$  ( $2.0$  to  $10 \text{ mM}$ ) to the equilibration medium containing  $150 \text{ mM } \text{K}^+$ . If this inhibition can be attributed to competition between a  $\text{TI}_{in}^+/\text{H}_{out}^+$  leak and the proton pump this effect may be seen as rapid entry of pump-depleted intravesicular  $\text{TI}^+$ . Hence the effect of  $2.0 \text{ mM } \text{ATP}$  on  $\text{TI}^+$  transport was assessed at  $6.0$  and  $0.6 \text{ mM } \text{TI}^+$  in the presence of  $150 \text{ mM } \text{KNO}_3$ . The higher  $\text{TI}^+$  concentration has been demonstrated to inhibit proton transport and the lower has no apparent effect under these conditions. In Fig. 4, trace *a*, the addition of ATP resulted in a rapid loss of intravesicular  $\text{TI}^+$  with a minimum achieved within  $10 \text{ sec}$  of substrate addition. Trapped  $\text{TI}^+$  at this point corresponded to  $30\%$  of the initial value. Similarly, ATP addition to the vesicles equilibrated in  $6.0 \text{ mM } \text{TI}^+$  plus  $150 \text{ mM } \text{K}^+$  also produced a rapid loss of trapped  $\text{TI}^+$ . In this case the trapped  $\text{TI}^+$  corresponded to  $45\%$  of initial levels. In  $6.0 \text{ mM } \text{TI}^+$  a rapid reuptake of ion occurred reaching  $90\%$  of initial levels within  $1 \text{ min}$  of ATP addition in contrast to that observed at the lower  $\text{TI}^+$  concentration. In both cases the quantity of ATP present at  $1 \text{ min}$  was sufficient to maintain transport; however, because of the requirement for intravesicular cation (either  $\text{K}^+$  or  $\text{TI}^+$ ) the rate of the transport reaction is slowed following initial cation depletion.

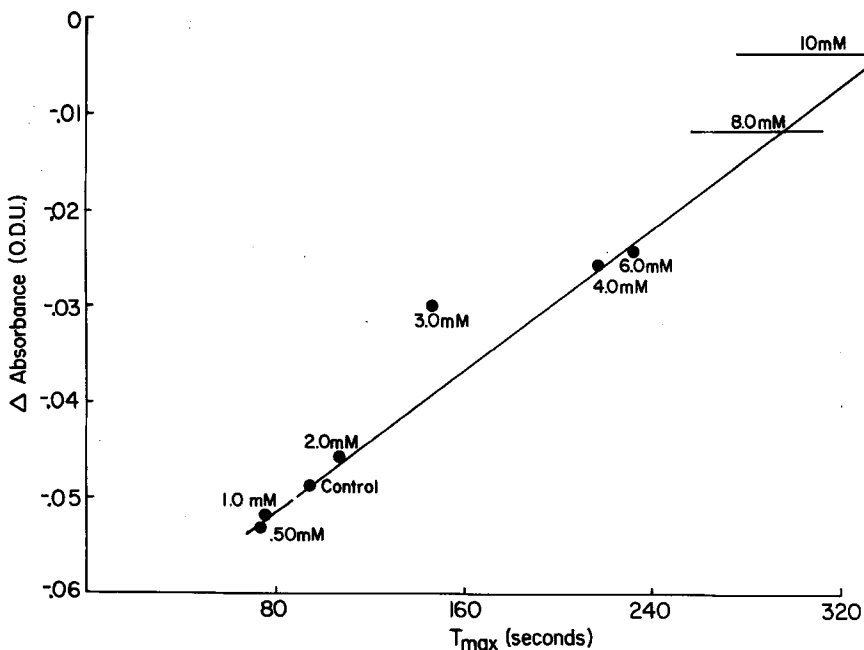
*Effect of  $Tl^+$  on the ATP-Dependent Proton Transport*

It has been shown previously (Sachs et al., 1976) that internal cation is required for ATP-dependent proton transport. This requirement may be met by the addition of valinomycin to nonequilibrated vesicles (in  $K^+$  or  $Rb^+$  media). In the case of  $Tl^+$ , in contrast to the equilibrium affinity of the valinomycin- $Tl^+$  complex which indicates some selectivity of valinomycin for  $Tl^+$  (Pressman, 1976), we do not observe a valinomycin-dependent component of  $Tl^+$ -stimulated ATP hydrolysis (results not shown). Since nigericin or gramicidin also dissipate proton gradients, preincubation is the only simple means available for allowing  $Tl^+$  entry. The activating effect of cations on ATPase activity is also complex. It has been suggested that whereas  $K^+$  binding to an internal cation site activates the ATPase,  $K^+$  binding to an external cation site inhibits the ATPase activity (Wallmark et al., 1980). At 2.0 mM ATP and 37 °C in medium at pH 7.4 this inhibition is seen at  $K^+$  concentrations of 8 mM or higher, and also starts to become evident above 6.0 mM  $Tl^+$  (Fig. 2).

When gastric vesicles were first pre-equilibrated with 50 mM  $TlNO_3$  at pH 6.1 and diluted 17-fold into 50 mM  $KNO_3$  or  $NaNO_3$  (not shown because of similarity to experiment resulting from  $KNO_3$  dilution) 30 sec prior to 0.5 mM ATP addition, the data of Fig. 5 were obtained. It was observed that under these conditions of high internal and low external  $Tl^+$  (50 mM  $Tl_{in}^+/2.6$  mM  $Tl_{out}^+$ ) acidification of the vesicles' interior occurs upon ATP addition. The additional



**Fig. 5.**  $Tl^+$ -dependent pH gradient formation. Thirty  $\mu$ l of  $Tl^+$  (50 mM) equilibrated gastric vesicles were diluted into 320  $\mu$ l of medium containing 50 mM  $KNO_3$ , 2.0 mM  $MgNO_3$  and 2.0 mM PIPES/Tris, pH 6.1. Transport was initiated within 30 sec of this mixing by the addition of 10  $\mu$ l of 30 mM MgATP. In a second experiment 5.0  $\mu$ l containing 0.1 mg/ml<sup>-1</sup> valinomycin was added approximately 50 sec after the addition of MgATP



**Fig. 6.**  $Tl^+$  inhibition of pH-gradient formation at pH 7.4. Gastric vesicles were equilibrated in a medium containing 150 mM  $KNO_3$ , 2.0 mM  $MgNO_3$ , 15  $\mu$ M acridine orange, 2.0 mM PIPES/Tris, pH 7.4, and variable  $Tl^+$  concentrations. The addition of 0.83 mM MgATP resulted in a decrease in the absorbance of acridine orange proportional to the magnitude of the pH gradient which is progressively inhibited by increasing  $Tl^+$  concentrations. Both the time of development ( $T_{max}$ ) of the minimum absorbance change and the absorbance change are plotted as a function of the equilibrated  $Tl^+$  concentration. Each point is the average of two assays

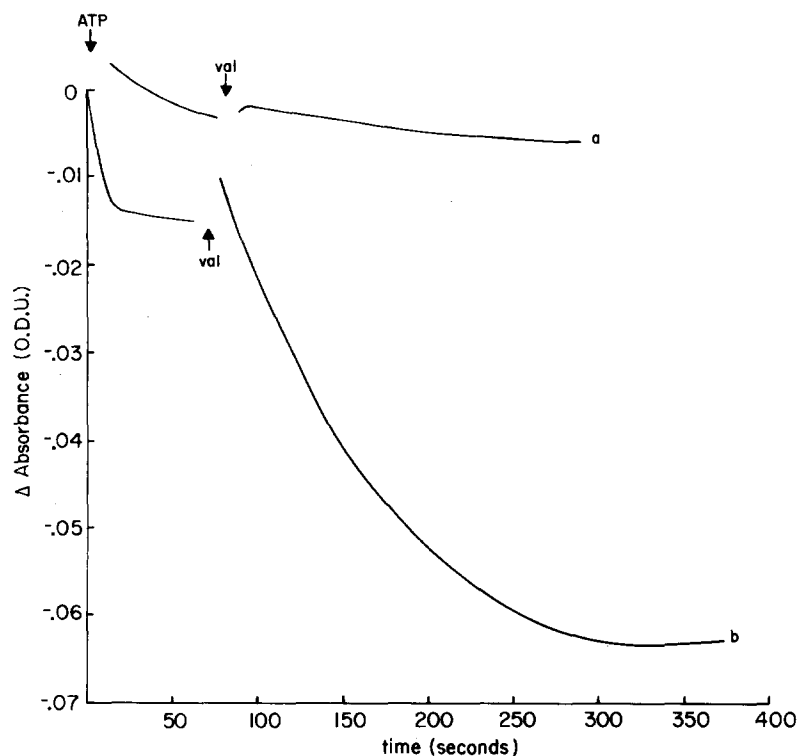


Fig. 7. Sidedness of  $Tl^+$  inhibition of proton pump activity. Gastric vesicles were equilibrated in a medium containing 150 mM  $KNO_3$ , 10 mM  $TlNO_3$ , 2.0 mM  $MgNO_3$  and 2.0 mM PIPES/Tris, pH 7.4. Immediately before the addition of 0.83 mM  $Mg^{++}$  ATP, 30  $\mu$ l of equilibrated material was diluted into 320  $\mu$ l of medium containing either 150 mM  $KNO_3$ , 2.0 mM  $MgNO_3$ , and 2.0 mM PIPES/Tris with or without 10 mM  $TlNO_3$ . Following the dilution MgATP was added and the absorbance change of acridine orange monitored. In each case, 5.0  $\mu$ l of 0.1 mg/ml valinomycin was added approximately 1 min after the addition of MgATP. Trace *a*: ATP-dependent formation of a proton gradient in material diluted into equimolar  $Tl^+$  (10 mM  $Tl_{in}^+$ /10 mM  $Tl_{out}^+$ ). Trace *b*: ATP-dependent formation of a proton gradient of material diluted into  $Tl^+$ -free medium (10 mM  $Tl_{in}^+$ /0.83 mM  $Tl_{out}^+$ ).

increase in acidification upon addition of valinomycin results from pump activity activated by the fresh supply of  $K^+$  supplied through the valinomycin  $K^+$  conductance pathway. If the vesicles are pre-equilibrated with 50 mM  $TlNO_3$  and ATP added directly without dilution, no proton uptake is observed.

To extend the above findings, the effect of  $Tl^+$  on  $K^+$ -dependent proton transport was measured in vesicles which were pre-equilibrated in a PIPES/Tris buffer, pH 7.4, containing 150 mM  $KNO_3$  and  $Tl^+$  at concentrations ranging from 0.5 to 10 mM as indicated in Fig. 6. In the vesicles equilibrated at 0.5 to 1 mM  $Tl^+$  ( $Tl_{in}^+ = Tl_{out}^+$ ) little effect on the rate of formation or magnitude of the proton gradient as compared to the absence of  $Tl^+$  is noted, other than a slight stimulation of the magnitude. In equilibrated medium containing higher  $Tl^+$  concentrations there is a progressive reduction in the rate of formation and magnitude of the proton gradient so that at 8.0 mM and above virtually no pH gradient was established in spite of the prior equilibration of  $K^+$  as well as  $Tl^+$ .

In order to establish the sidedness of the site responsible for  $Tl^+$  inhibition, vesicles equilibrated in 150 mM  $KNO_3$  and 10 mM  $TlNO_3$  at pH 7.4 were either diluted into identical solutions or into  $Tl^+$ -free solutions containing 150 mM  $KNO_3$ . Fig. 7 shows that in the equilibrated  $Tl^+$  medium (10 mM  $Tl_{in}^+$ /10 mM  $Tl_{out}^+$ ) the addition of ATP does not induce the formation of a pH gradient. However, when the external  $Tl^+$  concentration was decreased by dilution of the

$Tl^+$ - and  $K^+$ -equilibrated vesicles into  $Tl^+$ -free medium (10 mM  $Tl_{in}^+$ /0.83 mM  $Tl_{out}^+$ ), the formation of an ATP-dependent pH gradient was stimulated. Valinomycin had no effect on the formation of the pH gradient in the former conditions, but markedly stimulated the formation of the pH gradient under the latter.

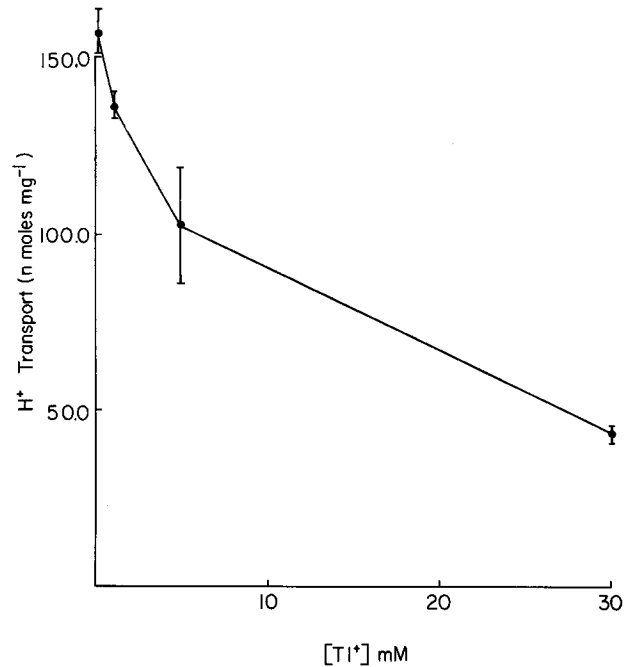
Since the ATP concentration used in these experiments is adequate to maintain effective substrate levels within the observed time limits, the valinomycin stimulation of pH-gradient formation can be ascribed to the ionophore-dependent supply of  $K^+$  to the cation pump site located in the vesicle interior (Ganser & Forte, 1973a). The absence of this effect in the presence of 10 mM  $Tl_{ext}^+$  indicates external rather than internal  $Tl^+$  inhibits the ATP-dependent proton pump.

It has previously been reported that the affinity of the external cation site is pH-sensitive (Stewart et al., 1980). Since the inhibition of the proton pump may in part be attributed to  $Tl^+$  binding at an external site it is possible in the intact vesicle system to test for this by measurement of the pH-dependence of  $Tl^+$ -induced inhibition. Accordingly, the inhibition resulting from ATP addition to vesicles equilibrated in medium containing 150 mM  $K^+$  and  $Tl^+$  in a range from 3.0 to 30 mM was measured at pH 6.1 in contrast to a media pH of 7.4 used previously. In this case, the pH electrode was used to measure proton pump activity since it is a direct measure of the proton component of the cation/ $H^+$  pump. In a control ex-

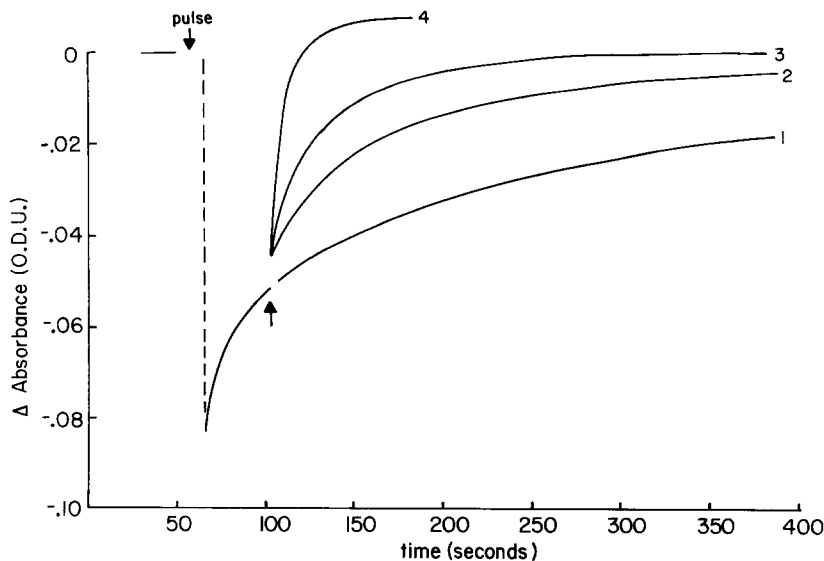
periment in which vesicles were equilibrated in medium containing 150 mM  $K^+$  and 10 mM  $Tl^+$  no ATP-dependent dye response was observed at pH 7.4 while only partial inhibition (approximately 25%) was seen at pH 6.1. In Fig. 8 progressive inhibition of net proton pump activity was again measured as a function of increasing  $Tl^+$  concentration. However, the degree of  $Tl^+$ -induced inhibition is reduced at pH 6.1 in comparison to pH 7.4. At 10 mM  $Tl^+$  only 40% inhibition of net proton uptake occurs in contrast to the almost total inhibition observed at pH 7.4. At 30 mM  $Tl^+$ , approximately 30% of the net transport measured in the absence of  $Tl^+$  is present at pH 6.1. Although in accordance with the biphasic  $Tl^+$  effect on the ATPase observed at pH 7.4 as well as the pH-sensitivity of the apparent affinity at the external cation site, the magnitude of this inhibition at pH 7.4 does not appear adequate to account for the complete loss of proton transport activity observed at concentrations of  $Tl^+$  above 8.0 mM. The inhibition of the formation of the pump-dependent pH gradient has an additional explanation.

#### Effect of External $Tl^+$ on Preformed pH Gradients

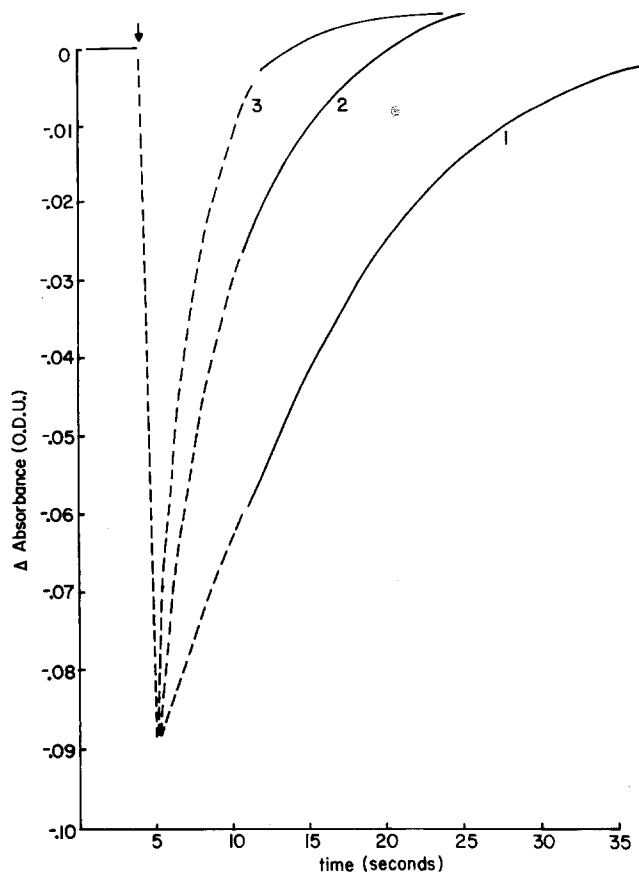
A pH gradient can be induced in gastric vesicles which are preloaded at pH 5.5 by the addition of external buffer at pH 8.5. The decay of the pH gradient in the absence of  $Tl^+$  is shown in trace 1 of Fig. 9. At 350 sec 80% of the pH gradient-dependent absor-



**Fig. 8.**  $Tl^+$  inhibition of proton transport at pH 6.1. Gastric vesicles were equilibrated in a medium containing 150 mM  $KNO_3$ , 2.0 mM  $MgNO_3$ , 1.0 mM PIPES/Tris, pH 6.1, and  $TlNO_3$  (1.0–30 mM). Each point is the average of duplicate assays which measure the maximum net levels of protons transported following the addition of 0.3 mM  $MgATP$  to each of the  $K^+$ - and  $Tl^+$ -equilibrated suspensions



**Fig. 9.**  $Tl^+$ -induced dissipation of a pH gradient in gastric vesicles. Gastric vesicles were preloaded with 150 mM  $KNO_3$ , 2.0 mM  $MgNO_3$  and 5.0 mM succinate/Tris, pH 5.5, as indicated in Materials and Methods. Thirty  $\mu$ l of the equilibrated material was diluted with 320  $\mu$ l of 150 mM  $KNO_3$ , 2.0 mM  $MgNO_3$ , 15  $\mu$ M acridine orange and 1.0 mM succinate/Tris, pH 5.5. At the indicated time 100  $\mu$ l of a solution containing 56 mM  $KNO_3$ , 2.0 mM  $MgNO_3$  and 15  $\mu$ M acridine orange and 100 mM Tris  $NO_3$ , pH 8.5, was added to the equilibrated material. Trace 1 shows the decay of the pH gradient in the absence of  $Tl^+$ . Trace 2 shows the decay of the pH gradient induced by the presence of 3.8 mM  $Tl^+$ . Trace 3 shows the decay of the pH gradient induced by the presence of 8.0 mM  $Tl^+$ . Trace 4 shows the decay of the pH gradient induced by the addition of 0.5  $\mu$ g nigericin



**Fig. 10.**  $Tl^+$ -induced dissipation of a pH gradient in liposomes. Liposomal vesicles were prepared as indicated in Materials and Methods. Ten  $\mu$ l of the equilibrated vesicles were then diluted into a 340- $\mu$ l solution containing 276 mM mannitol, 2.0 mM  $MgNO_3$ , 15  $\mu$ M acridine orange and 1.0 mM succinate/Tris, pH 5.5. At the indicated time a 100- $\mu$ l solution containing 100 mM Tris  $NO_3$ , pH 8.5, 176 mM mannitol, 2.0 mM  $MgNO_3$  and 15  $\mu$ M acridine orange was added to the suspension containing liposomes. Trace 1 shows the pH-gradient decay in the absence of  $Tl^+$ . Trace 2 shows the decay of the pH gradient in the presence of 0.8 mM  $Tl^+$ . Trace 3 shows the decay of the pH gradient in the presence of 4.0 mM  $Tl^+$

bance response has dissipated. In the following experiments shown in traces 2 and 3, 4.0 and 8.0 mM  $Tl^+$  (final) were added to the external medium during the slow decay phase. These  $Tl^+$  additions resulted in a concentration-dependent decay of the pH gradient. The addition of the  $H^+/K^+$  exchange ionophore, nigericin, produced the rapid decay of the pH gradient shown in trace 4. The effect of external  $Tl^+$  on the pulse-induced pH gradient in these gastric vesicles can be ascribed to the presence of a  $Tl^+ : H^+$  exchange (or  $Tl^+ : OH^-$  co-transport).

To distinguish between a  $Tl^+$  interaction between membrane proteins or lipid, pH gradients were induced in a similar manner across protein-free pH 5.5 mannitol-loaded liposomes formed from phosphatidyl choline. In Fig. 10 the dissipation of the pulse-

induced pH gradient is indicated in trace 1. In subsequent experiments 0.8 and 4.0 mM  $Tl^+$  were introduced into the external medium with the pulse solution. As in the gastric vesicles the decay of the pH gradient was dependent on the concentration of externally added  $Tl^+$ . If the lipid permeable cation triphenylmethylphosphonium ( $TPMP^+$ ) is substituted for external  $Tl^+$  an equivalent dissipation of the pH gradient formed in gastric vesicles is obtained at 1 mM cation. This suggests a finite  $H^+$  conductance is present in these vesicles and makes plausible the interpretation that the  $Tl^+ / H^+$  exchange is electrically coupled.

### Discussion

The gastric ATPase which usually exchanges  $H^+$  for  $K^+$  is able to use  $Tl^+$  as a substitute for  $K^+$ .  $Tl^+$  activates the ATPase with a relatively high affinity in comparison to  $K^+$ . The gramicidin enhancement of  $Tl^+$ -stimulated hydrolysis is indicative of an intravesicular location of the  $Tl^+$ -activating site from which  $Tl^+$  is depleted by the activity of the  $H^+ / Tl^+$  pump upon addition of substrate. Location of the  $Tl^+$ -stimulating site on the internal face of the membrane is also consistent with the protection from the inhibition of hydrolysis of the ATPase by the carboxyl group reagent ethoxycarbonyl ethoxydihydroquinoline by the presence of intravesicular  $K^+$  (Saccomani, Barcellona, Rabon & Sachs, 1980).

The  $Tl^+$  concentrations required for inhibition of gramicidin-stimulated ATP hydrolysis is at least two orders of magnitude greater than that required for activation. A biphasic effect has also been reported with  $K^+$  with inhibition of maximal hydrolysis beginning at approximately the same range as observed with  $Tl^+$  (Wallmark et al., 1980). With  $K^+$  it was shown that inhibition resulted from inhibition of phosphorylation by bound external  $K^+$ . The inhibition by  $Tl^+$  due to binding at an external site is strongly suggested by the antagonism between medium pH and  $Tl^+$  inhibition observed in the measurements of proton transport, as well as by the  $Tl^+$  dilution experiments.

As with  $Rb^+$  (Schackmann et al., 1977) or  $K^+$  (Lewin, Saccomani, Schackmann & Sachs, 1977)  $Tl^+$  is transported by the ATPase. At low  $Tl^+$  concentrations, ATP-induced  $Tl^+$  efflux and the outward  $Tl^+$  gradient is maintained.  $Tl^+$  influx due to lipid permeation of the ion is considerably faster than that of the alkali metals  $K^+$  and  $Rb^+$  and at high  $Tl^+$  concentrations this permeability presumably has the effect of causing  $Tl^+$  efflux to be transient, efflux being followed by reuptake.

With pH gradients imposed across the membrane



in the absence of ATP,  $Tl^+$  addition results in dissipation of the gradients.  $Tl^+$  probably also dissipates the pH gradient due to the ATPase by a similar  $Tl^+/H^+$  exchange. This interpretation is supported by the  $Tl^+$ -induced dissipation of pH gradients in protein-free liposome, and a similar effect of TPMP<sup>+</sup> in the gastric vesicles.

Thus the net effect of  $Tl^+$  on the ATPase is determined by cation binding at the catalytic subunit with either the internal activating or external inhibitory cation-binding site. In parallel to the pump pathway for  $Tl^+$  there is a  $Tl^+/H^+$  exchange pathway which accounts for the uncoupling of ATPase activity from flux of cation and  $H^+$ , in the presence of external  $Tl^+$ .

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